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# ODOR REDUCTION THANKS TO TEXTILE MATERIALS

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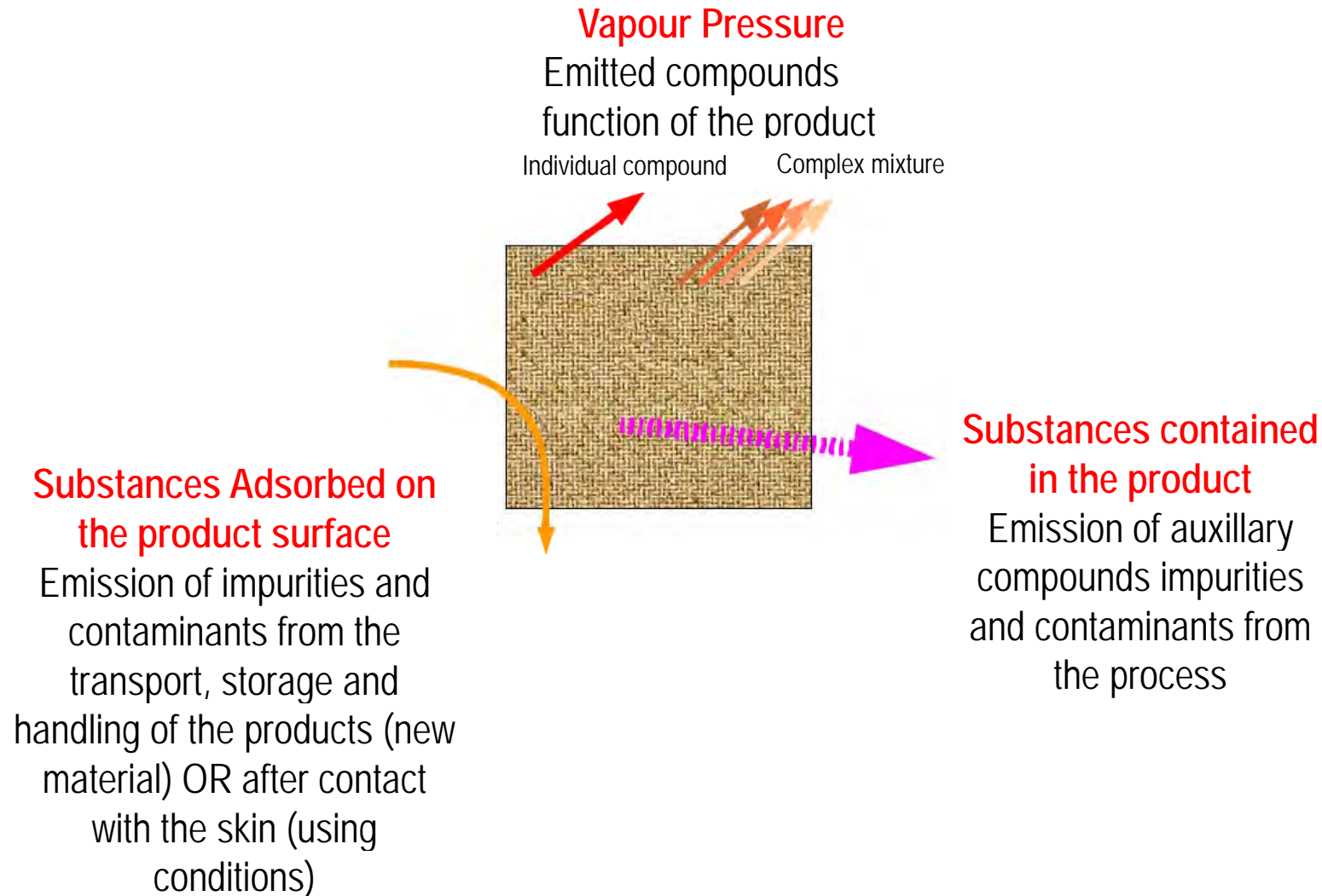
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6<sup>th</sup> RESET Seminar on  
“New materials and new applications”  
Huddersfield, 31.01.2018

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## Origin of odours from Textiles



## Textile and skin interactions



- SKIN = not sterile surface : Ecosystem which plays an important role in the body balance
- Resident micro-organisms : various bacteria adapted to the physiological skin conditions : surface temperature between 30°C and 35°C, pH from 5 to 6,5, various nutritive substances (perspiration, sebum from the sebaceous glands, cellular fragments)
- Major skin microflora strains : Staphylococcus (*Staphylococcus aureus* and *epidermidis* – Gram-positive), micrococcus, aerobic and anaerobic corynebacteria, propionobacteria and, in case of lack of hygiene : Gram-negative bacteria
- Transient micro-organisms (Temporary colonization) : constituted of many micro-organisms from endogenous (from the body) or exogenous (environnement) sources.

## Textile and Bacteria interactions



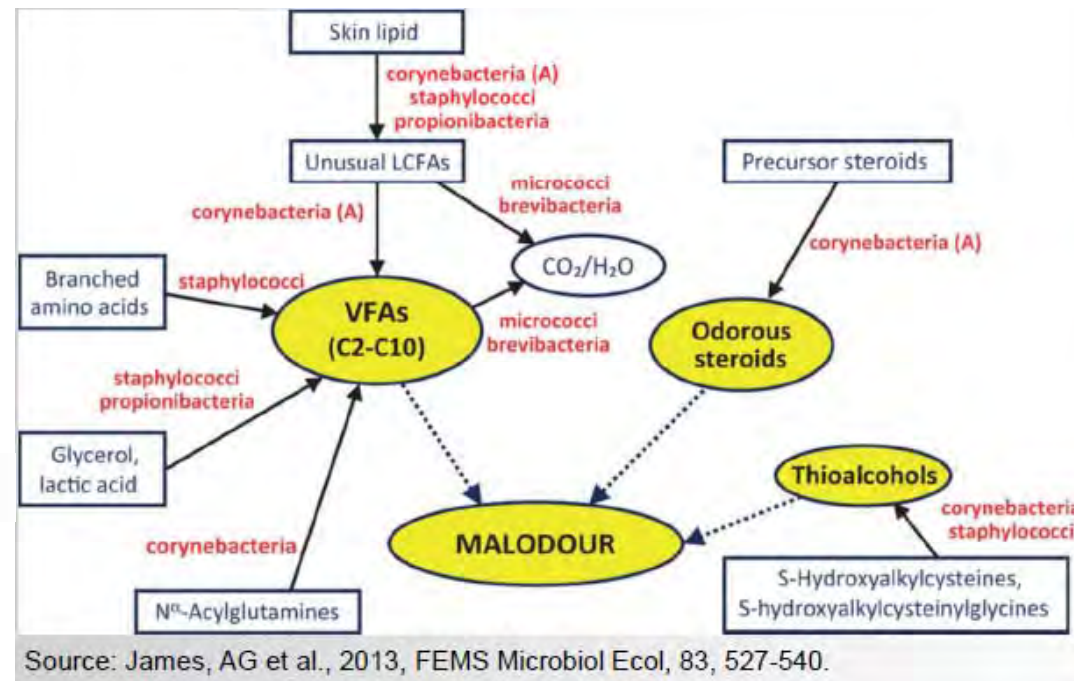
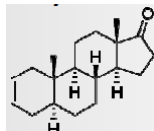
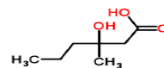
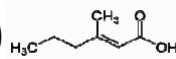
- **Role of the textile in odour generation** : Act as a barrier which block the water evaporation and increase its condensation. The consequent moisture combined with the presence of nutrients at the interface between the skin and the textile induce the growth of micro-organisms and the potential to generate malodour.
- Reinforced by the high specific surface of the textile products
- After 8H of « normal » wearing, the bacteria population reach  $10^4 / \text{cm}^2$
- The body odour intensity is considered as a malodour from  $10^7$  germs/cm<sup>2</sup>. As the number of bacteria can double about every 30 minutes, a malodour can be perceptible after 12 hours of a « normal » wearing.

## Interaction Textile / Odorous substances

- Relation between malodour and number of total specific bacteria **Coryneforms** (*James et al. 2004*)
- Other bacteria participate to the global sweat odour : **Staphylococci** and **Micrococci**

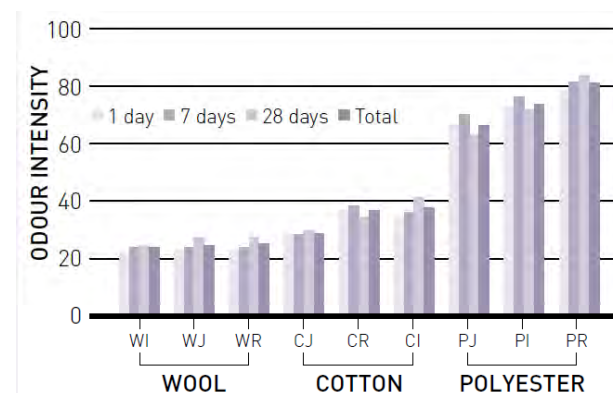
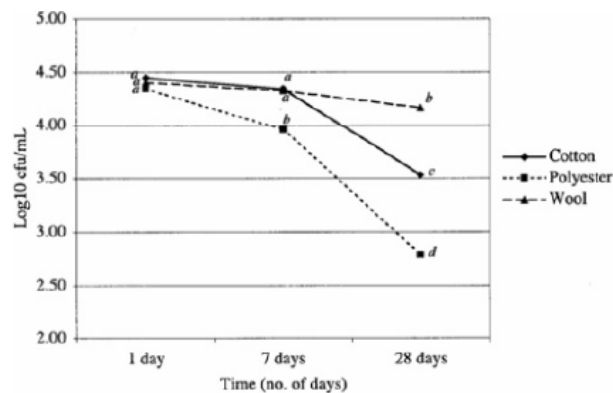
- Odorous compounds issued from the bacteria degradations :

- ✓ 3-Methyl-2-hexenoic acid (axillary)
- ✓ Isovaleric acid (feet)
- ✓ 3-hydroxy-3-methylhexanoic acid
- ✓ Thioalcohol
- ✓ Androstenone
- ✓ ...



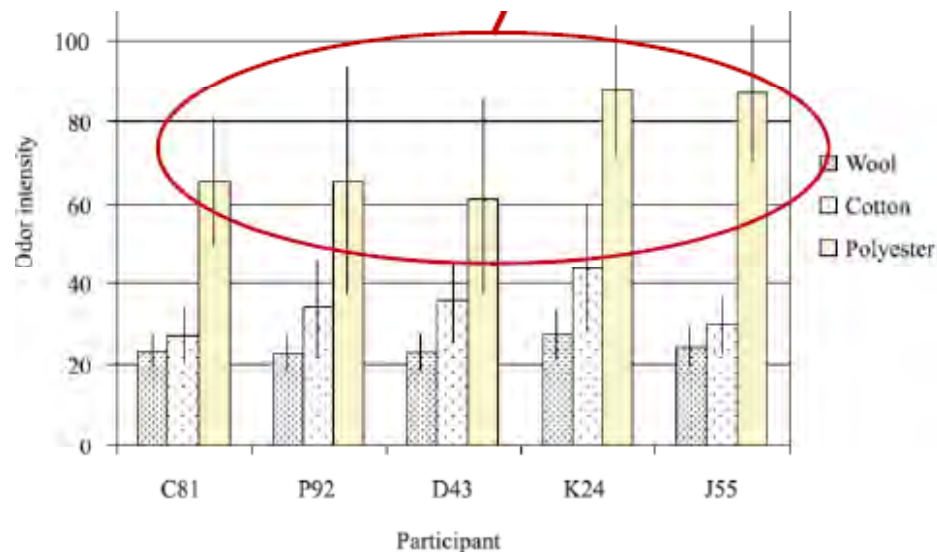
## Textile – Bacteria interactions in the odour release

- First Observations :
  - ✓ Higher adhesion of bacteria on hydrophobic, non polar surfaces : Polyester vs. Cotton (*Fletcher, M., 1996. Bacterial adhesion*)
  - ✓ Lower growth of Staphylococcus on hydrophilic compared to hydrophobic fibers (*Teufel and Redl, 2006*)
- Complementary Studies :
  - ✓ Comparable microbial numbers on wool, cotton and polyester after 1 day use (*Teufel and Redl, 2006*)
  - ✓ Higher survival rate of bacteria on wool than on polyester or cotton
  - ✓ BUT : odour intensity is lower on wool



Source : Mc. Queen et al. (2007). Odour Intensity on apparel fabrics and the link with bacterial populations. *Textile Research Journal*, 77, 449-456

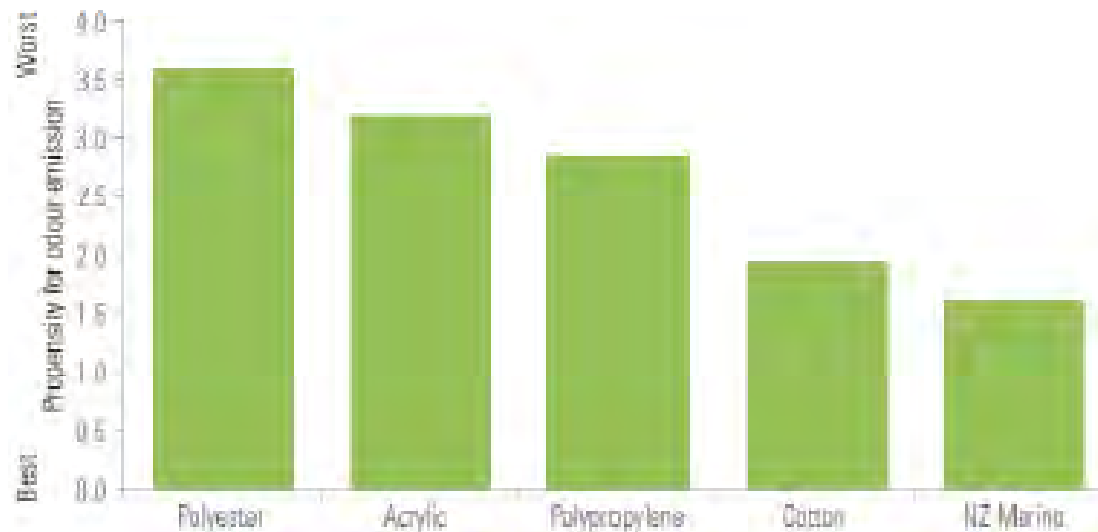
## Odour retention on fabrics – Axillary odours



- The odour intensity is higher for hydrophobic fibers regardless of sweat composition (Polyester >> Cotton and Wool)
- Significant results were obtained in particular for Wool fabrics which are less odorous after wear than polyester (panel of 13 assessors)
- Short-chain carboxylic acids were detected as responsible of the polyester odour after wear (Otago University)
- An other study (New Zealand) found that wool fabrics retained about 66% less body odour intensity than polyester fabrics and 28% less than cotton fabrics (olfactory measurement)

## Odour retention on fabrics – Feet odours

- Sensory Study on odours emissions from socks made with different fabrics
- From The Wool Research Organisation New Zealand Inc. (WRONZ), become Canesis Network Ltd and then AgResearch Ltd.



- Confirm previous studies (Wool, far less odours compared to Polyester or Acrylic)

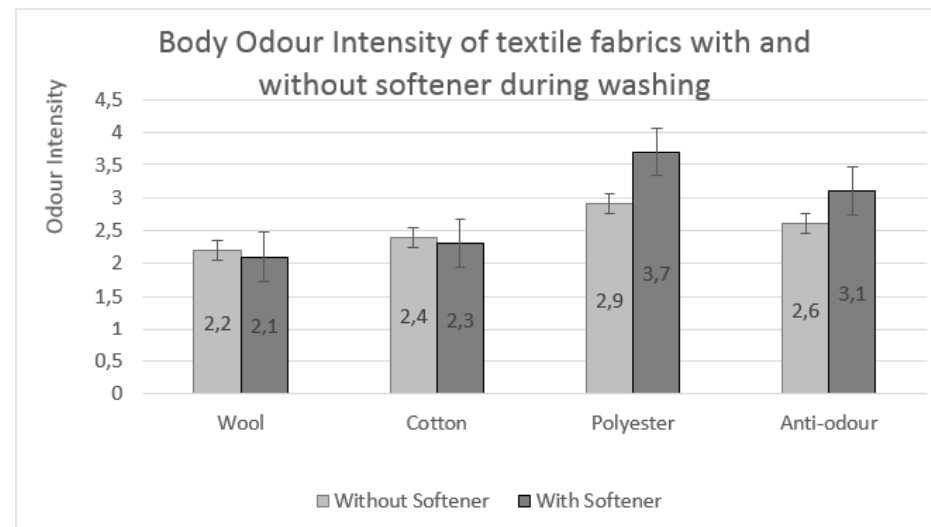


## Synthesis : Influence of the type of fiber on bacteria and odour

	Polyester	Cotton	Viscose	Acrylic	Polyamide	Wool
Odour	+++ (fatty acids, Ketones and aromatic comp.)	+ (Aldehydes)	+	++	++	- / + (Aldehydes / Ketones)
Bacteria initial development	++	+	+	+	+	+
Bacteria persistancy	++ (2, 4, 5)	+ (1, 2, 5)	-- (1, 3, 4)	- (1, 2, 3, 4)	+ / - (1, 5) (2, 3)	++ (1, 2, 4, 5)
<i>Origin</i>	Develop. of specific odour releasing strains					

## Impact of washing on textile odours

- Fabric softeners seems to enhance the bad odour on Polyester (*Laitala et al. 2012*)
- Lipase in detergents may have an impact on odour formation (Munk et al. 2000)
- Volatile substances like carboxylic acids are faster removed in a washing process than micro-organisms (Chung and Seok, 2012)
- Volatils substances are easier removed on Cotton than on Polyester (McQueen et al., 2013)
- Volatiles from washing machine contribute to laundry malodour (Stapleton et al., 2013)



Source : *Laitala et al. (2012). Troubles with the Solution : Fabric Softeners and Odour Properties. Tenside Surf. Det., 49 (5), 362-368.*

## Potentially odorant substances at the interface of human skin and textile

\* Typical Sweat Odours

Odour

**S**

H<sub>2</sub>S Mercaptans\*  
Sulfides\*  
Thiazoles  
Thiophenes

**N**

Amines\*  
Pyrazines\*  
Pyrridines  
Nitriles  
Pyrroles  
Indole

Odour

**O**

Alcohols\*  
Aldehydes\*  
Ketones / Esters\*  
Acids\*  
Furanes  
Acetals

Odour

Odour

**Hydrocarbons**

Alkenes  
Terpenes  
Some Alkanes



**Aromatics**

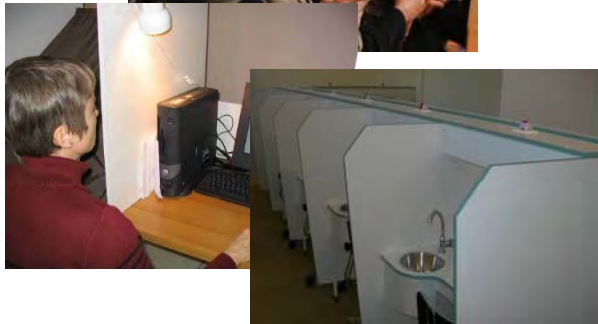
BTEXS / BHT

Odour

## Testing

### → Olfactory Evaluation :

- External parameters control (French norms (sensory analysis) AFNOR V09 105, SSHA).
- Samples management
- Measurement of Detection, Intensity, type of Odours (+ deodorization efficiency, following ISO 17 299)
- Repeatability and reproducibility of the measurement



### → Microbiological testing :

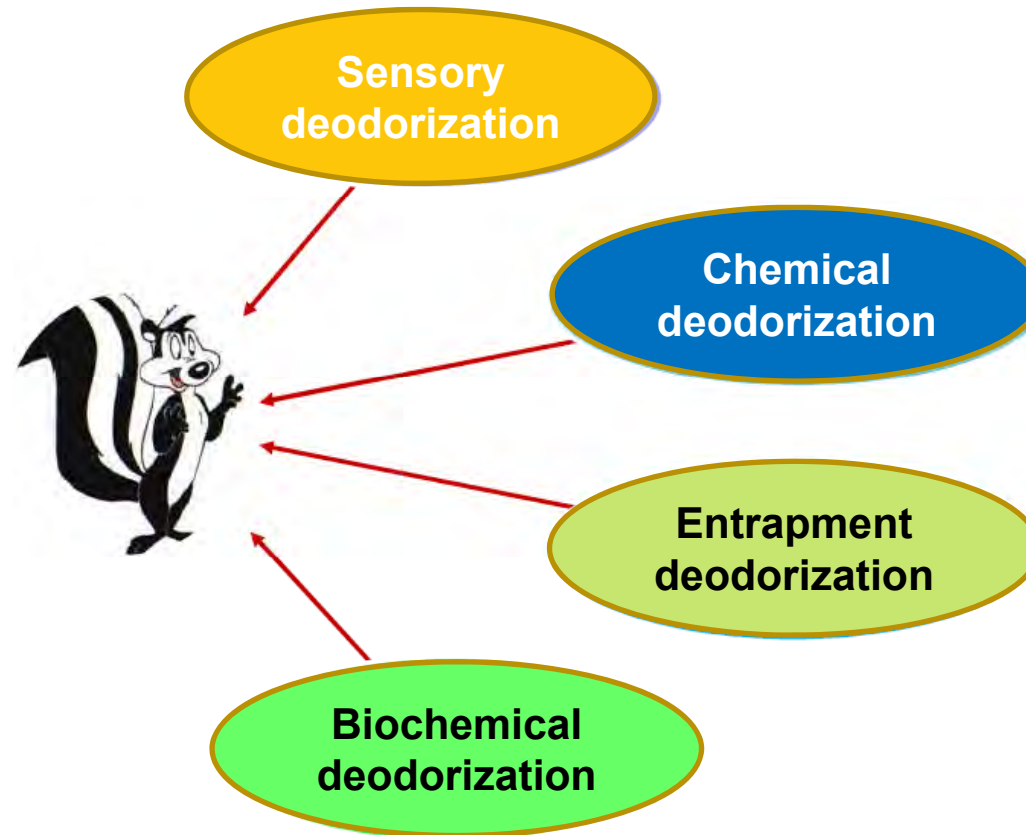
- ISO 20743, ASTM2149, JIS L 1902 & AATCC100



### → Other Tests :

- Durability (Wash, Abrasion)
- VOC (GC/MS and HPLC/MS)
- Air or water permeability

## Odour elimination in textiles



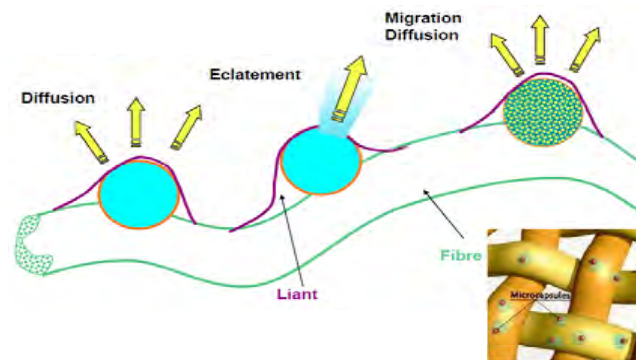
## Odour elimination in textiles

- Sensory deodorization :

*Textile : controled release (microcapsules, resines)*

- Masking Agents (biology / biochemistry) : Hiding bad odours (essential oil) : Competitive chemical bonding with the olfactory receptor (« fragrance finishing ») : masquodor® (Protex)
- Neutralizing Agents (chemitry) : Based on a chemical reaction directly with the malodourous substance (air or liquid phase) : Decrease of the « bad » odour intensity or modification of the chemical structure (odour modificalton)

- Processes : Impregnation (crosslinking agent), Spraying, Coating, Incorporation in the fibre.
- Stage : Impregnation : thermal fixation (=at 130 to 170°C (drying and curing step)
- Release : Physical (dialytic : wall insoluble and product soluble in water and body fluids) / Biochemical (enzymatic degradaton) / Chemical (dissolution by specific reactions)



*Microencapsulation*

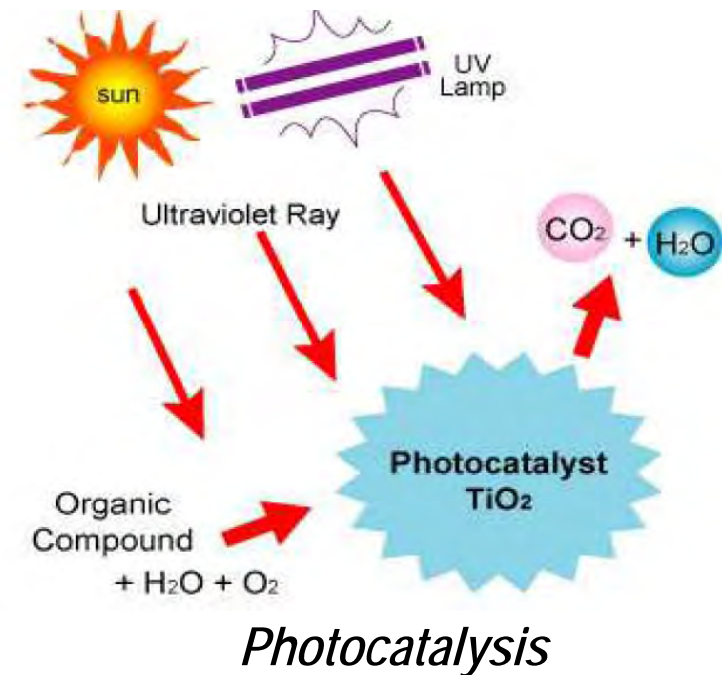
- Advantages : many manufacturing processes / Several types of textiles
- Disadvantages : Curing temperature decreased aroma retained inside the microcapsule / washing durability (< 25 home launderings) / Biocidal products regulation (BPR) 528/2012

## Odour elimination in textiles

### - Chemical deodorization :

*Textile : Addition of chemically active products to degrade odorous substances (ex : TiO<sub>2</sub> for photocatalysis)*

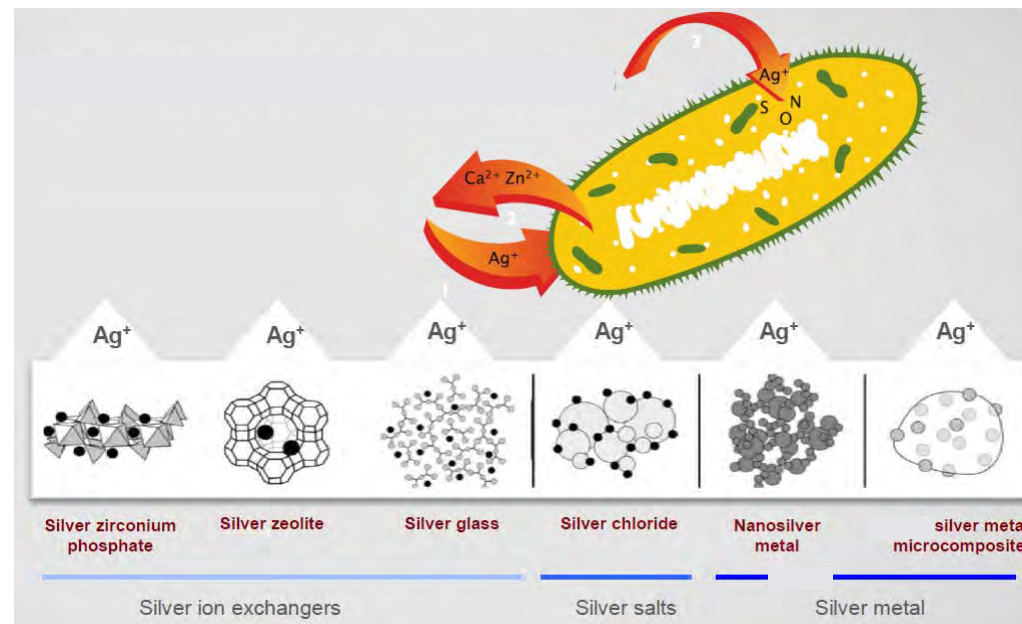
- Chemical reaction to transform substances with a bad odour in other substances without odour or with a better odour
  - Redox réactions (organic compounds),
  - Acido-basic (nitrogenous compounds with acidic reagent or sulfur compounds with basic reagent)
  - Ionic exchange (ammoniac or sulfure compounds neutralization by ferric salts or acétaldéhyde, H<sub>2</sub>S and amines by polyoxometalates)
  - Photocatalysis (Reaction between radical species from O<sub>2</sub> and H<sub>2</sub>O and organic compounds)



## Odour elimination in textiles

- Biochemical deodorization:  
*Textile : biocides*

- Bactériostatic : Hinder proliferation of bacteria (without changing bacteria's skin flora).
- Bactericide : Eliminate micro-organisms (lethal action)





## Odour elimination in textiles

- Biochemical deodorization:

*Textile : biocides*

- Regulations (biocide directive N°98/8/CE : Biocide market regulation)
- Selection of the technology depending on the objective (fabric protection or deodorization) and on the manufacturing process (melting process : dissolution in the melting polymere (PES, PA et PP), Dry or Solvent process : dissolution in the solvent (cellulosic acetate, acrylic and chlorofibers) or wet process: dissolution in water (viscose)

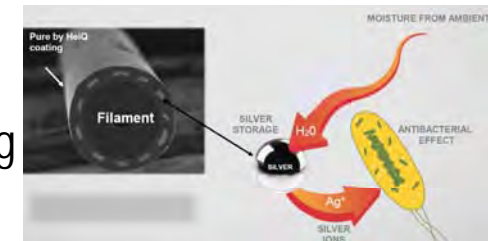
Biocide	Ratio (%)	Price (\$/kg)
Silver	32,5	70* - 130
Silane Quats	13	30 - 50
Quats	4,5	50

*Source : Report Biocides in Textiles, 2017*

And... Triclosan, Zinc Pyrithione, Izothiazolinones (BIT, MIT, CMIT), formaldehyde releasers, phenolic derivates, paraben, copper, organic acids, alcohols

## Odour elimination in textiles

- **Biochemical deodorization**: formulation and application processes
  - Biocide coated finishes (90% production in volume of pure product)
    - Aegis, Aglon, Purista (Lonza), Irguard (BASF), Foamfresh (Piedmont), Sanitized, Ultrafresh (Thomson), Silvadur (Dow), Pure TF (HeiQ)
    - Processes : Exhaustion, Padding, Spraying
    - Stage : Pre-spinning (on thread), Finishing (on raw textile), Spraying
  - Biocide Inherent fibres (10% production)
    - Allerban (Advansa), Coolmax (Asota), Saniguard (Miroglio), Rhovyl, Bactershield (Sinterama), Mushon (Toray), Bioactive (Trevira), Radilon (Radici), XT2 and X-static Fibers (Silverescent products, Noble), Newlife (Polygiene / Sinterama)
    - Processes : Synthetic fibre production
    - Stage : Spinning solution, mixed at the melt polymer



### Limits :

- Some studies did not prove the efficiency of antibacterial treatments on odour reduction (*McQueen et al. (2013), Journal of the textile institute, 104 (1), p. 108*) and the durability is still called into question (Swedish Chemicals Agency, KEMI, PM 8/15, 2015)
- High cost of treatment

# Odour elimination in textiles

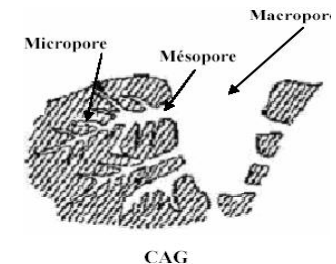
- Entrapment deodorization :

*Textile : Trapping charges (physical and chemical interactions : cyclodextrines, **activated carbons**, silica gel, argiles, zeolites)*

## ACTIVATED CARBON

- Odorous substances are chemically or physically fixed on active sites at the surface of the material  
Hydrogen, Van der Waals or ionic bonds (+ chemical functionalisation to increase reactivity)
- Chemical selectivity depending on :
  - ✓ The pore size (macropores : from 50 to 2000 nm, mesopores : from 2 to 50 nm, micropores : < 2 nm)
  - ✓ The polarity (hydrophilic characteristics)

*Activated  
Carbon*



## NEW DEVELOPMENTS

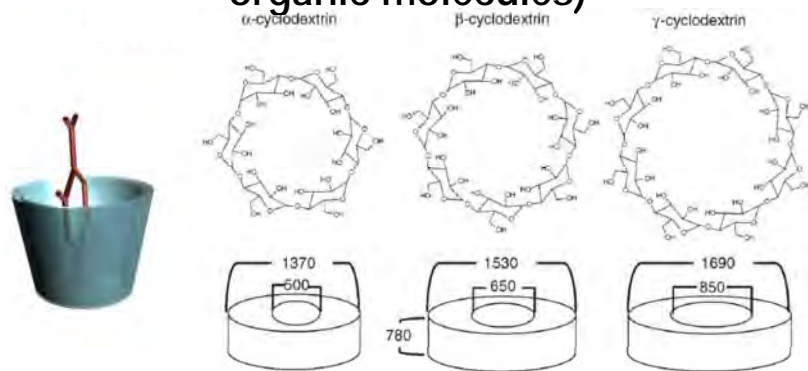
- Activated carbon fibers (in comparison with standard activated carbon functionalisation : larger specific surface, better adsorption capacity, better accessibility to micropores, faster adsorption kinetics)
- Porous polymers: Control of the specific surface and hydrophilic parameters, purity

Products : HeiQ Fresh NKU (zeolites),  
Scentry (Microban, activated carbons),

# Odour elimination in textiles

- Entrapment deodorization - CYCLODEXTRINES

Torus-shaped molecules with hydrophobic cavities between 0,5 and 0,85 nm (trapping of organic molecules)



Advantage : **Cyclodextrines**

- Do not interact with dyeing products (colour and odour efficiency)
- Efficiency proved on real samples (cigarette smoke: *Setthayanond, J et al. (2017). Cellulose, vol.24, issue 11, pp.5233-5250*)

Table 1. Feasible interactions between  $\beta$ -CD and some textile fibres [14].

Parameter	Cotton	Wool	PES	PA	PAN	PP
Ionic interactions	-	+	-	+	+	-
Covalent bonds	+	+	-	+	-	-
Van der Waal forces	-	-	+	+	+	-
Crosslinking agents	+	+	+	-	-	-
Graft polymerisation	+	+	+	+	+	+

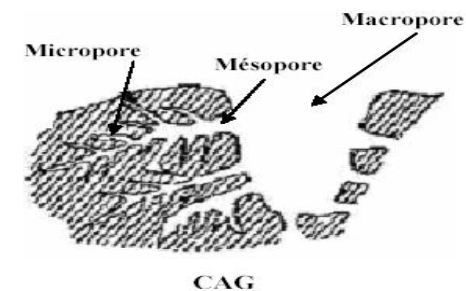
+=possible,-=not possible, PES-polyester, PA-polyamide, PAN-polyacrylonitrile, PP-polypropylene.

*AOBCI. AUTEX Research Journal, vol.11, N°4; Dec 2011*

## Anti-odour fabrics – Market Study

### ➤ Entrapment technology

- Carbon based (Activated carbon, activated carbon fibers...)
  - Zorflex (Calgon)
  - Cocona (replaced by 37,5)
  - Scafé
  - Bamboo charcoal (Acelon Chemical)
  - Saratech (Blücher)
- Minerals (zeolites, clays, nanomaterials...)
  - LAVA XL (Sciessent)
- Polyméric (Synthetic polymers...)
- Molecular (Cyclodextrines)



Selectivity  
Efficiency

## Anti-odour fabrics

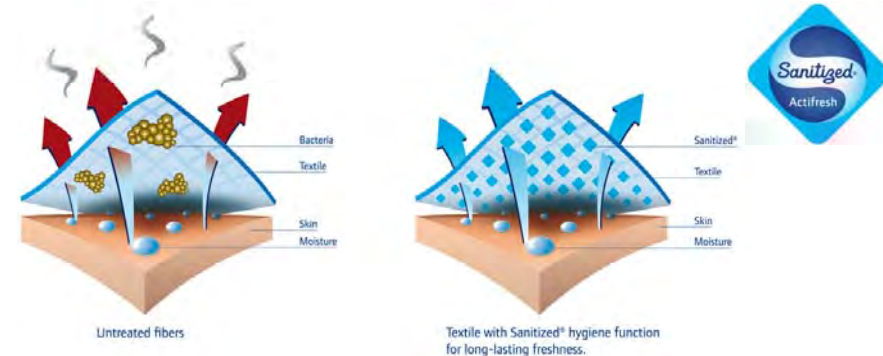
- Activated Carbon Fibers



Standard Claims
Ammonia, Trimethylamine
Methylmercaptan, Hydrogensulfide, Methyldisulfide
Carboxylic acids (C2 to C4), Aromatic compounds, Acetic ether, Methyl isobutyl Ketone, i-butanol
Aldehydes (C3 to C5)

## Anti-odour fabrics

- Cyclodextrine solutions
- Pulcra Chemicals GmbH : Anti-odour treatment Cyclofresh® (+ silver : cyclofresh plus®)
  - o Liquid application
  - o Body odours absorption + Fragrance release
  - o Suit with many types of fabrics
  - o Regenerated by washing or spraying
- Sanitized AG : Traitement ACTIFRESH®
  - o Liquid / Solid applications (liquid, paste, masterbatch, powder), for extrusion, padding, extraction, spray, coatings



## Anti-odour fabrics

- Photocatalytic Fibers

**SELCLEAR** is a self-cleaning fiber that uses light as a catalyst

**Exlan, Japan**

Photocatalytic reaction (like  
reactions to sunlight)  
and their effects is conveyed  
throughout the fiber due to the  
incorporated nanovoids (spaces).

Nanovoid

Photocatalytic  
nano titanium  
oxide

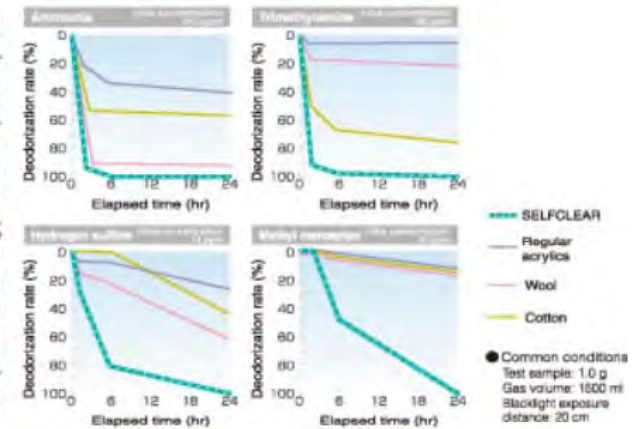


### Effect on 4 main odors

4 main troublesome odors

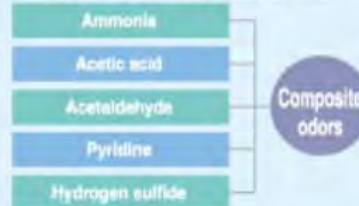
Causal substances	Odor	Source
Ammonia	Urine-like odor	Urine or sweat
Trimethylamine	Rotting fish-like odor	Rotting fish or swab
Hydrogen sulfide	Rotting egg-like odor	Rotting eggs, excrement, hairbois, or raw garbage
Methyl mercaptan	Rotting onion-like odor	Rotting onions, excrement, hairbois, or raw garbage

Exceptional deodorizing performance proven by data

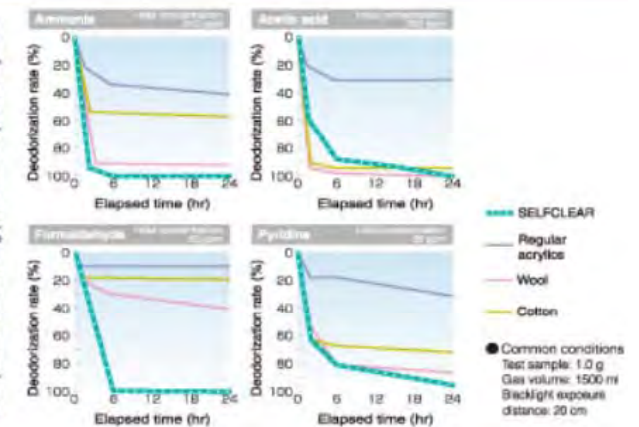


### Effect on tobacco odor

Tobacco odors are composite odors created from the following substances



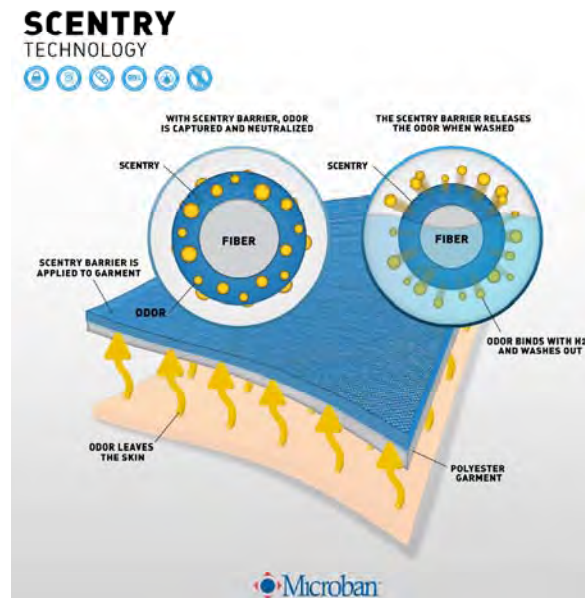
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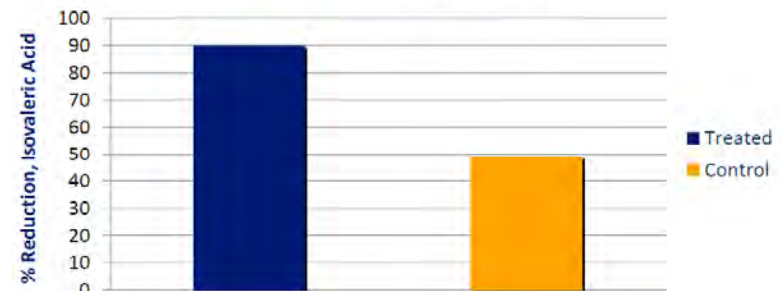


## Anti-odour fabrics

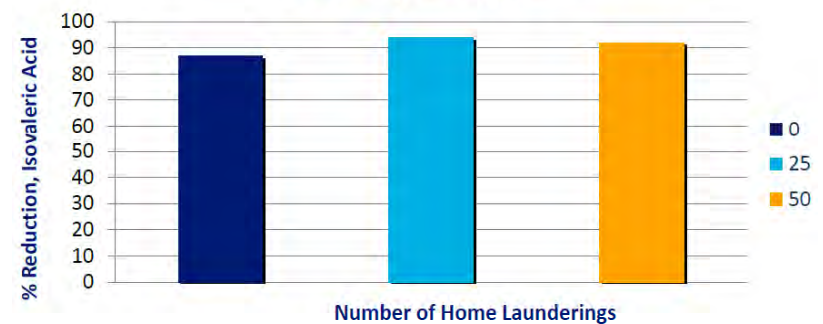
- Antibacterial agent
  - ✓ Liquid-based coating, applied after dyeing (during finishing process)



Odor Capture Performance of 5% Scentry-treated Polyester vs. Untreated Polyester



Durability of Odor Capture to 50 HL: 5% Scentry, 0.3% ZPTech



Source : Welch, K.T.; Lan, T. & Aylward, B. Scentry (2016). Scentry® and ZPTech® : A bifunctional, Effective, and Durable Odor Solution for Polyester Activewear



# RESET

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**Thank you!**

